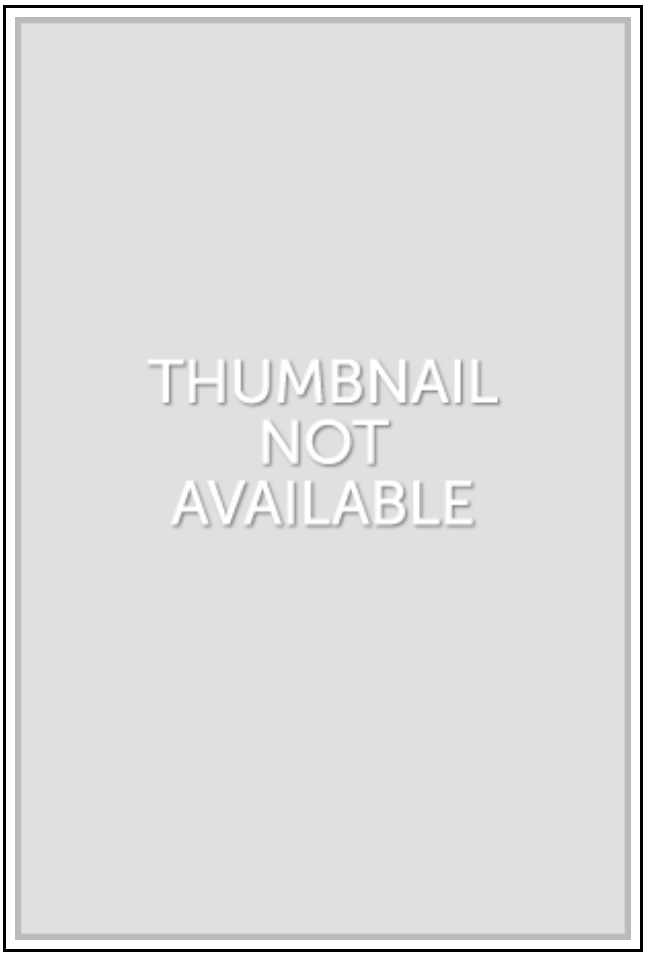
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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1899 Excerpt: .and careful observation and measurement; this, however, is not infallible, and apparent violations of the foregoing statements will be encountered, though they may be generally used with consistent results. 2567. At ordinary temperatures and pressures, five of the elements, hydrogen, oxygen, nitrogen, fluorine, and chlorine, are gases; mercury and bromine are liquids; while all the rest, including all metals excepting mercury, are solids. All the solids except carbon have been liquefied at various temperatures. Few elements, except the more common metals and oxygen and nitrogen, are extensively used or found in an uncombined state; they usually occur in various combinations, which are divided into three classes: acids, bases, and salts. 2568. An acid may be defined as a compound containing hydrogen, which hydrogen may be replaced by a metal when presented to it in the form of an oxide or hydrate. The combination of oxygen with any other single element is called an oxide. A bydrate is the substance formed by the union of an element, or, more often, a metallic oxide, with the elements of water. This should not be confounded with a solution, which is merely a mechanical mixture of some solid with water, or similar liquid. Acids are usually sour to the taste, and will readily cause chemical action. Most acids contain oxygen, being formed from the union of an oxide of a non-metal and water; but in some few acids oxygen is absent. Table 90 gives a list of the more common...

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